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Solid solution, mixture of two crystalline solids that coexist as a new crystalline solid, or crystal lattice. The mixing can be accomplished by combining the two solids when they have been melted into liquids at high temperatures and then cooling the result to form the new solid or by depositing vapours of the starting materials onto substrates to form thin films.

Solid solution | chemistry | Britannica

CBSE Class 12 Chemistry Lab Manual Introduction to Basic Laboratory Equipment Viva Questions with Answers. Surface Chemistry Exp-2.1 : To prepare colloidal solution (sol) of starch. Exp-2.2 : To prepare a colloidal solution of gum. Exp-2.3 : To prepare colloidal solution (or sol) of egg albumin. Exp-2.4 : To prepare ferric hydroxide, $[\text{Fe}(\text{OH})_3]$ sol. Exp-2.5 : To prepare aluminium hydroxide, [...]

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Identify each of the underlined items as a part of either the macroscopic domain, the microscopic domain, or the symbolic domain of chemistry. For any in the symbolic domain, indicate whether they are symbols for a macroscopic or a microscopic feature. (a) The mass of a lead pipe is 14 lb. (b) The mass of a certain chlorine atom is 35 amu.

Ch. 1 Exercises - Chemistry 2e | OpenStax

Organic chemistry employs solid-liquid, liquid-liquid, and acid-base extractions. The following applies to liquid-liquid extractions, which will be used in this course. It is very common for organic products synthesized in a reaction to be purified by liquid-liquid extraction. ... Place the solution to be extracted in the separatory funnel. As ...

Chem 211 - Techniques

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