

## Stoichiometry Stumper 1 Answer

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### Stoichiometry Stumper 1 Answer

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### stoichiometry stumper #1 by Amy Hinds - Prezi

Stoichiometry Stumper #1 You are a forensic scientist. You are investigating a murder involving poison. The victim was poisoned with a compound called di-chloro benzene whose formula is C6H4Cl2.

### Solved: Stoichiometry Stumper #1 You Are A Forensic Scientist ...

Stoichiometry Stumper #1 You are a forensic scientist. You are investigating a murder involving poison. The victim was poisoned with a compound called di-chloro benzene whose formula is C 6 H 4 Cl 2. Autopsy results show that the victim's body contained about 31 g of the poison, but the actual amount could have been slightly

### Stoichiometry Stumper #1 - Tumwater School District

Stoichiometry Stumper 1 Answer Forensic Author: electionsdev.calmatters.org-2020-10-22T00:00:00+00:01 Subject: Stoichiometry Stumper 1 Answer Forensic Keywords: stoichiometry, stumper, 1, answer, forensic Created Date: 10/22/2020 5:50:48 PM

### Stoichiometry Stumper 1 Answer Forensic

Stumper 1 Answer Thank you for reading stoichiometry stumper 1 answer. I'm trying to study for a test but these three keep stumping me... The answers are 1. 359.9 g/mol 2. C4H7 3. 164g. Can you explain how to get those? 1.Cortisone consists of molecules each of which contains 21 atoms of carbon. The mass percentage of carbon in cortisone is 69.98% . What is the molecular mass of cortisone? 2.A 2.00-g sample of a ...

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Stoichiometry Worksheets with Answer Keys August 6, 2020 Some of the worksheets below are Stoichiometry Worksheets with Answer Keys, definition of stoichiometry with tons of interesting examples and exercises involving with step by step solutions with several colorful illustrations and diagrams.

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3 1 1.47 30.0 g 122.55KCl 1 mol KClO mol KCl 3 g KCl 1 3KCl mol KClOmol = g KClO49.3 74.55 2 2.73 g 1KClO 3 mol KCl molg KClO KCl 3 2 mol KCl O 3 = g KCl 122.55 1 1.66 74.55 g2 KCl KCl 42.7 g Fe mole Fe mol2 Fe 2O 3 g Fe Femol FeO 3 g Fe 2O 3 = g Fe 2O 3 55.85 4 1 61.0 17.0 g O 2 1 mol 2O 2 mol Fe 2O 3 g O32.00 3 mol O 2 g Fe 2O 3 g Fe 2O 3 = 1 ...

### stoichiometry 1 worksheet and key - Saddleback College

This was a problem in my chemistry review packet. It's the only one I don't know I don't know how to do. I've been working on it for hours. A mixture of methane (CH4) and ethane (C2H6) is burned. The two combustion reactions require a total of 179.959 grams of oxygen, and they produce a total of 226.473 grams of products. From this information, calculate the masses of methane and ethane in the ...

### Stoichiometry Stumper? | Yahoo Answers

1 mol O x mol H O 22.4 L 2O 2 2 2 2 2 D. Find the mass of tristearin required to produce 55.56 moles of water (about 1 liter of liquid water). 57 110 6 57 110 6 57 110 6 2 57 110 6 57 110 6 2 899g C H O 1 mol C H O 890 g C H O 110 mol H O 2 mol C H O x 55.6 mol Answers: 254A. 9.9 x 10 atoms Mn 4C. 33.2 mol Mn 3 O 4 5A. 1168 L O 2 5C. 0.675 ...

### Stoichiometry: Problem Sheet 1

The Answer Our company does not have enough funds to produce the calcium nitrate needed to make the required amount of the final drug. Our company will have to sell the rights to the drug to a more well-funded company because we only have \$40,000 and the amount needed is about

### Stoichiometry Stumper #3 by Jadine Lee - Prezi

Stoichiometry Worksheet #1 Answers 1. Given the following equation: 2 C 4H 10 + 13 O 2--> 8 CO 2 + 10 H 2O, show what the following molar ratios should be. a. C 4H 10 / O 2 b. O 2 / CO 2 c. O 2 / H 2O d. C 4H 10 / CO 2 e. C 4H 10 / H 2O 2. Given the following equation: 2 KClO 3--> 2 KCl + 3 O 2 a. How many moles of O 2 can be produced by ...

### Stoichiometry Worksheet #1 Answers - PSD401

Stoichiometry Level 1, Level 2, Level 3 Answers. Basic Skills in Chemistry Workbook Stoichiometry Level 1, Level 2, Level 3 Answers: Level 1: 1. 55.7 g 2. .982 g 3. 164 g 4. 64.2 g 5. 19.4 g. Found: 12 Mar 2020 | Rating: 84/100. Practice Test Ch 3 Stoichiometry Name Per. Remember it is a MC test, use the answers ...

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